

RESEARCH ARTICLE OPEN ACCESS

Prediction of Fragmentation Pathway of Natural Products, Antibiotics, and Pesticides by ChemFrag

Jördis-Ann Schüler¹  | Annemarie E. Kramell² | Antonia Schmidt¹ | Pauline D. Walesch¹ | René Csuk²

¹Institute of Computer Science, Martin Luther University Halle-Wittenberg, Halle (Saale), Germany | ²Department of Organic Chemistry, Martin Luther University Halle-Wittenberg, Halle (Saale), Germany

Correspondence: Jördis-Ann Schüler (joerdis-ann.schueler@informatik.uni-halle.de)

Received: 29 July 2024 | **Revised:** 11 March 2025 | **Accepted:** 18 March 2025

Funding: This work was supported by the German Research Foundation (425225219).

Keywords: fragment ion annotation | mass spectrometry | natural products | rule-based fragmentation | semiempirical quantum mechanics

ABSTRACT

Because the manual interpretation of ESI-MSⁿ fragmentation spectra is time-consuming and usually requires expert knowledge, automated annotation is often sought. The fragmentation software ChemFrag enables the annotation of MSⁿ spectra by combining a rule-based fragmentation and a semiempirical quantum chemical approach. In this study, the rule set was extended by 31 cleavage rules and 12 rearrangement rules and used for the interpretation of ESI(+)-MSⁿ spectra of antibiotics, pesticides, and natural products as well as their structural analogs. The fragmentation pathways predicted by ChemFrag for compounds such as 17 β -estradiol were confirmed by a comparison with pathways published in other studies. In addition, the annotations were compared with those of the programs MetFrag and CFM-ID, for example, with regard to the number and intensity of annotated fragment ions. Our experiments show that ChemFrag provides reliable and in some cases chemically more realistic annotations for the fragment ions of the investigated compounds. Thus, ChemFrag is a helpful addition to the established in silico methods for the interpretation of ESI(+)-MSⁿ spectra.

1 | Introduction

Mass spectrometry (MS) with their various configurations is unquestionably a powerful tool for the identification and quantification of organic compounds. Depending on the sample introduction system, ionization technique, and mass analyzer, organic compounds with very different properties can be characterized and various issues addressed. Electrospray ionization (ESI)-MS in combination with liquid chromatography (LC-ESI-MS or LC-ESI-MS/MS) is used for numerous applications, for example, in medicine or biochemistry. This method is suitable for the characterization of nonvolatile, thermally unstable, or polar compounds even in complex matrices. In addition to other techniques such as nuclear magnetic resonance (NMR) or infrared (IR) spectroscopy, fragmentation (MS² or MSⁿ) spectra in particular can contribute to structure elucidation. Manual interpretation of fragmentation

spectra is time-consuming and usually requires expert knowledge. Thus, an automated interpretation of the generated data, which enables high-throughput screening, is helpful for many applications. One approach for the identification of unknown compounds is the comparison with spectral libraries [1] such as MassBank, currently containing 96.350 MS² spectra, or the NIST and Wiley databases. However, spectral databases based on ESI-MS/MS are relatively small. The comparison of spectra is further complicated by the fact that device parameters such as collision energy are often decisive for the number and intensity of fragment ion peaks. Another approach for the annotation of MS/MS spectra is the application of in silico methods [2]. These include rule-based fragmentation (e.g., Mass Frontier HighChem, Ltd. Bratislava, Slovakia, versions after 5.0 available from Thermo Scientific, Waltham, USA) [3], combinatorial fragmentation [4, 5], comparison of fragmentation trees [6, 7], or machine-learning based approaches [8–10]. Established

This is an open access article under the terms of the [Creative Commons Attribution](https://creativecommons.org/licenses/by/4.0/) License, which permits use, distribution and reproduction in any medium, provided the original work is properly cited.

© 2025 The Author(s). *Journal of Mass Spectrometry* published by John Wiley & Sons Ltd.

programs are, for example, MetFrag [11], CFM-ID [12], or SIRIUS [13]. These methods were originally developed for the identification of metabolites based on the measured spectrum. To identify the metabolite, they determine, among other things, the structures of the fragment ions. The advantage of these programs is their short runtime; however, chemically unfeasible fragment ions are occasionally generated. In contrast, quantum chemical-based methods [14], such as QCMS² [15, 16] or QC-FPT [17], form chemically correct fragment ions. However, in many cases, they require a significantly longer runtime to do so. In addition, the required software is usually not available free of charge. Approaches based on quantum chemistry as well as rule-based fragmentation are combined in the program ChemFrag [18]. With its shorter runtime than established quantum chemical methods and its chemically more plausible annotation than rule-based fragmentations, ChemFrag provides the basis for the prediction of fragmentation pathways of different classes of organic substances. Furthermore, the semiempirical PM7 method implemented in the Molecular Orbital PACkage program (MOPAC), which is available to users free of charge in an academic context, is used for the quantum chemical calculations.

In this study, we present the further development of ChemFrag for the interpretation of ESI(+)-MSⁿ fragmentation spectra of antibiotics, pesticides, natural products, and their structural analogs such as steroids, flavonoids, sulfonamides, and butanoic acid esters. This includes the implementation of new rules and the verification of predicted fragmentation paths. Some fragmentation pathways are compared with those already published, whereas for other compounds, fragmentation ions are annotated for the first time.

2 | Materials and Methods

2.1 | ChemFrag Architecture/Workflow

ChemFrag is a tool for the interpretation of MSⁿ fragmentation spectra that combines a rule-based approach with quantum chemical calculations (see [18]). For the quantum chemical calculations, the semiempirical PM7 method is used, which is implemented in MOPAC. This method provides heats of formation, which are used to select chemically meaningful fragment ions for subsequent fragmentation steps, as well as bond orders for identifying weak bonds. Using the rule-based approach, ChemFrag generates energetically stable fragment ions. ChemFrag currently incorporates 44 cleavage rules and 16 rearrangement rules, derived from the literature [19–27] or quantum chemical computations.

In the following, we will briefly describe the process of fragmentation simulation using ChemFrag. The first step involves the ionization of the input molecule (positive ion mode: protonated molecule ions [M+H]⁺). Subsequently, stable molecule ions are selected for the initial fragmentation step on the basis of reaction enthalpies calculated by the heat of formations of the products and reactants. Next, fragment ions are generated by applying cleavage and rearrangement rules and by cleaving bonds with a low bond order. The chemical plausibility of the formed fragment ions is evaluated using semiempirical calculations. Chemically meaningful fragment ions are selected for the next fragmentation step. This cycle continues until a specified termination criterion, for example, the fragmentation depth, is met.

If an m/z value of a generated (fragment) ion matches a signal in the experimentally determined MSⁿ spectrum, the (fragment) ion is assigned to this peak. ChemFrag's output includes the assignment of molecule and fragment ions to the experimentally determined m/z values as well as a fragmentation tree. The fragmentation tree allows the reconstruction of reaction pathways leading to the formation of fragment ions. The implemented cleavage and rearrangement rules can be extended dynamically. To simulate various ionization strengths, users can adjust parameters such as reaction enthalpies or fragmentation depth. Moreover, the integration of other semiempirical quantum chemical methods such as the GFN2-xTB method [28] into ChemFrag is conceivable. However, this is the subject of future studies.

2.2 | Implementation of new Fragmentation Rules

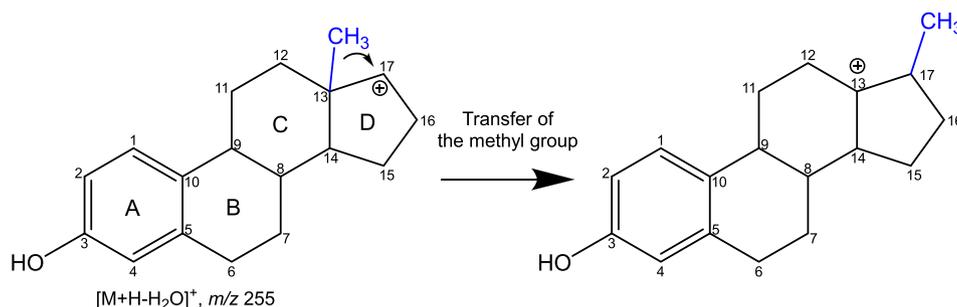
Compared with the first version of ChemFrag, which was used for the interpretation of ESI(+) CID mass spectra of doping substances such as ephedrine and cocaine, the rule set has been expanded [18]. The cleavage and rearrangement rules implemented in the current ChemFrag version are shown in Tables S1 and S2 as well as in [18]. The newly implemented rules allow the investigation of further compound classes. One of the newly introduced rules is the methyl shift, which enables a chemically meaningful annotation of MS/MS spectra of protonated steroids ([M+H]⁺). The migration of methyl residues according to a Wagner–Meerwein rearrangement was reported for 10-, 13-, and 17-methyl steroids such as methyltestosterone, methandienone, 5 α -androst-1-en-17 β -ol-3-one, estrone, 17 α -estradiol, and 17 β -estradiol (1) [20, 29–31]. As shown in Scheme 1 for 1, a positive charge at C-17 triggers the 1,2-transfer of the methyl group from C-13 to C-17, forming a stable tertiary carbocation.

ChemFrag uses a partial structure of the steroid system, consisting of the C- and D-ring, to check the applicability of the methyl shift. In order to keep the rule more general, a six-membered ring is also permitted instead of the five-membered ring. Specifically, ChemFrag now checks on the basis of the SMARTS [CH3]C12CCCC1CC[C+]2 and [CH3]C12CCCC1CC[C+]2 whether a suitable substructure is contained in the molecule. As soon as the substructure test is successful, ChemFrag conducts the following steps for the rearrangement (for the numbering of the C atoms see rearrangement rule 13; Table S2):

1. The bond between the C-1 atom and the methyl group is cleaved.
2. A new single bond is formed between the methyl group and the charged C-9 atom at the five-ring or C-10 at the six-ring.
3. The charge of the positively charged atom is reduced by one.
4. The charge of the C-1 atom is increased by one.

2.3 | ESI-MSⁿ Analyses

ESI-MSⁿ spectra were recorded using a Finnigan LCQ mass spectrometer (ion source: ESI, cation sensitive detection, spray



SCHEME 1 | Migration of a methyl residue starting from fragment ion [M+H-H₂O]⁺ of **1** ESI(+) rearrangement rule: methyl shift for steroids.

gas: N₂, damping and collision gas: He, CID mass spectrometry). Two acquisition modes were used to characterize each compound: full scan mode and MSⁿ of the precursor ion. The steroids have been provided by hapila GmbH (Gera, Germany); all other compounds have been a gift by Organica GmbH (Bitterfeld-Wolfen, Germany).

3 | Results and Discussion

To evaluate the evolution of ChemFrag, the program was applied to 22 compounds from different substance classes. To verify the effectiveness of the new rules, the experiments were divided into several sections. First the fragmentation pathway of estrogenic steroid **1** as predicted by ChemFrag was compared with a previously published one. Second, the number of annotated fragment ions from ChemFrag was compared with those obtained by established programs such as MetFrag and CFM-ID. In addition, the newly implemented rules were used to predict the fragmentation behavior of several steroids, such as estriol 3-methyl ether (**2**) and Δ^{9,11}-dehydro-17α-cyanomethylestradiol (**3**). For the latter compounds, the authors are not aware of any ESI(+)-MSⁿ fragmentation spectra with annotated fragmentation ions. Consequently, the fragmentation paths of these compounds are predicted for the first time in this study. Finally, other classes of compounds were included in the analysis, and mass spectra of compounds such as 2-cyano-2-phenylbutanoic acid ethyl ester (**4**), nicotinamide (**5**), and the flavonoid quercetin (**6**) were interpreted and annotated by ChemFrag.

3.1 | Annotation of ESI-MSⁿ Spectra of Various Steroids and Comparison to MetFrag and CFM-ID

In the first experiment, the reaction pathway predicted by ChemFrag for protonated **1** (**E1**, [C₁₈H₂₅O₂]⁺, *m/z* 273.19, Scheme 2) was compared with that of Ma and Yates of 2018 [20], which closely follows the results of Bourcier et al. of 2010 [31]. In both cases, water elimination is predicted after protonation of the hydroxyl group at C-17, forming a secondary carbocation (**E2**, [C₁₈H₂₃O]⁺, *m/z* 255.17). To generate a more energetically stable tertiary carbocation (**E3**), ChemFrag and Ma and Yates both propose a 1,2-transfer of the methyl group from C-13 to C-17, followed by an opening of the C-ring. The opening of the C-ring forms a primary carbocation, which is converted into a tertiary carbocation by hydride transfer (**E4**). This is followed

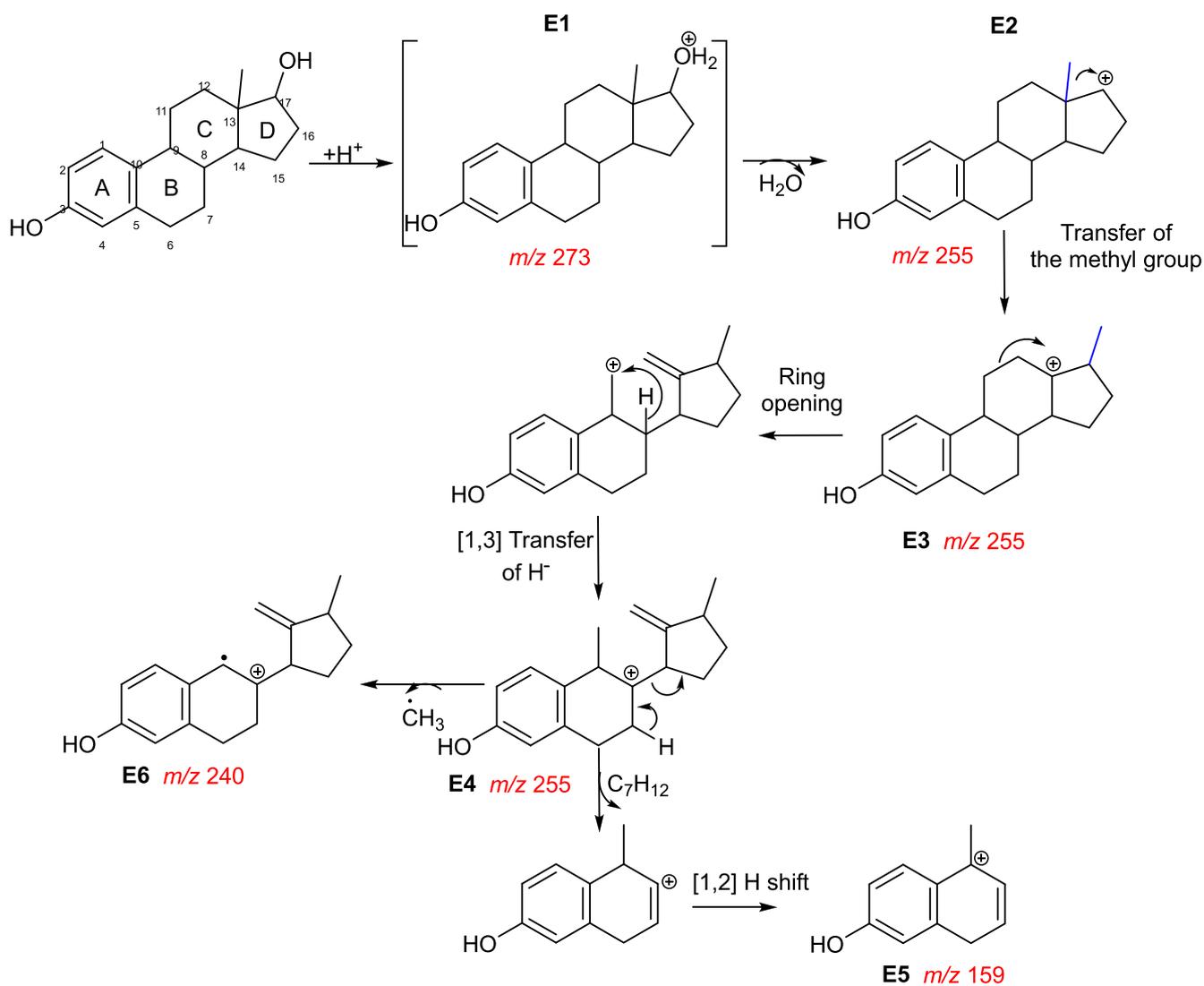
by the cleavage of the D-ring, a rearrangement of the positive charge to the benzylic position and the formation of the resonantly stabilized naphthalenol derivative **E5** at *m/z* 159.08 ([C₁₁H₁₁O]⁺). In contrast to Ma and Yates, ChemFrag additionally predicts the loss of a methyl radical from the benzylic position of **E4** and the formation of the stable distonic radical cation **E6** at *m/z* 240.15 ([C₁₇H₂₀O]⁺). The loss of an alkyl radical from [M+H]⁺ ions is known for steroids and has already been described by Guan et al. [32]. In addition, LC-ESI(+)-MS² spectra of compound **1** available in the MassBank database also show a peak at *m/z* 240.15 [MS analyzer: quadrupole time-of-flight (Q-TOF); collision energy (CE): 20–50 eV] [33].

In summary, the newly implemented rules for the rearrangement of the methyl group and the opening of the six-membered ring lead to a reaction pathway that is consistent with the literature. Thus, we have demonstrated that ChemFrag is capable of predicting chemically meaningful annotations and is suitable for the prediction of steroid fragmentation pathways.

Moreover, ChemFrag was tested for other steroids, and the annotation rates of the total of nine steroids were compared with those of MetFrag 2.0 and CFM-ID 4.0 (Table 1, for structures see Figure S1). In this experiment, the absolute score and the weighted score were used. The absolute score is the number of annotated peaks out of the total number of peaks in the spectrum (peaks with an rel. intensity ≥ 5% or 10% are taken into account in the evaluation; see Tables 1 and 4). The weighted score takes into account the rel. intensity, that is, the annotation of peaks with high intensity has a stronger influence than that of low intensity peaks. The weighted score (*s*) is calculated using the Formula (1), which forms the sum over the set of intensities of the annotated peaks (*F*). The total weighted score of a spectrum is the sum over all peak intensities. While the absolute score shows how many peaks of a spectrum we can explain, the weighted score improves mainly when high intensity peaks, not necessarily more peaks, are annotated.

$$s = \sum_{f \in F} \text{intensity}_f \quad (1)$$

As shown in Table 1, ChemFrag annotates the most (fragment) ion peaks with higher intensity. In contrast, the total weighted score of MetFrag for the examined steroids is about 600 lower, which is probably related to the missing annotation of the protonated molecule ion ([M+H]⁺). In contrast to ChemFrag and CFM-ID, MetFrag only annotates fragment ion peaks of the



SCHEME 2 | Fragmentation pathway of protonated **1** $[M+H]^+$ predicted by ChemFrag [(fragment) ions described by Ma and Yates [20]: m/z 273, 255, 159].

TABLE 1 | Comparison of the weighted scores and the absolute scores (in brackets) of selected steroids. The annotation of the (fragment) ions was performed with ChemFrag, MetFrag 2.0, and CFM-ID 4.0 (scores in the format “determined score/maximum score”; peaks with a rel. intensity $\geq 5\%$ are taken into account in the evaluation).

Substance	ChemFrag	MetFrag	CFM-ID
17 β -Estradiol (1)	220/220 (4/4)	120/220 (3/4)	210/220 (3/4)
Equilin 3-acetate	180/188 (4/5)	166/188 (4/5)	14/188 (1/5)
Estriol 3-methyl ether (2)	305/390 (9/16)	34/390 (4/16)	100/390 (1/16)
9(11)-Dehydroestrone	134/234 (4/5)	140/234 (4/5)	125/234(3/5)
Estriol 3-acetate	119/227 (4/6)	181/227 (4/6)	38/227 (1/6)
$\Delta^{9,11}$ -Dehydro-17 α -cyanomethyl-estradiol (3)	321/343 (5/8)	199/343 (3/8)	100/343 (1/8)
17 α -Cyanomethyl-estradiol	285/325 (5/9)	213/325 (6/9)	100/325 (1/9)
Estriol 17-acetate	164/164 (4/4)	152/164 (3/4)	12/164 (1/4)
α -Hydroxyestrone diacetate	205/219 (5/7)	158/219 (5/7)	55/219 (1/7)
Total score	1933/2310(43/64)	1363/2310 (36/64)	754/2310 (13/64)

respective compound but not the signal of the $[M+H]^+$ ion. If the signal of the protonated molecule shows a high intensity, this nonexplanation is clearly reflected in the score. In comparison, the total weighted score of ChemFrag is almost three times as high as that of CFM-ID. The absolute scores of CFM-ID show that mostly only one or at most three fragment ions of the steroids could be annotated. We therefore conclude from this experiment that the number of fragment ions annotated by ChemFrag is comparable to MetFrag and is higher than CFM-ID. A comparison of the generated structures also shows that ChemFrag

achieves chemically more meaningful results than CFM-ID. For example, in the case of **1** ($C_{18}H_{24}O_2$), CFM-ID generates a fragment ion with a protonated hydroxyl group at C-3 after the elimination of CH_4 and H_2 ($[C_{17}H_{19}O_2]^+$, m/z 255.14, Table 2). In comparison, as described above, ChemFrag predicts a loss of water after protonation of the hydroxyl group at C-17, followed by a 1.2 methyl shift, opening of the C-ring and hydride transfer, forming the fragment ion **E4** ($[C_{18}H_{23}O]^+$, m/z 255.17; see Scheme 2). As described above, the formation of fragment **E4** is consistent with the studies of Ma and Yates [20].

TABLE 2 | Comparison of selected fragment ion structures of **1**, generated by ChemFrag and CFM-ID.

ChemFrag		CFM-ID	
m/z	Fragment ion	m/z	Fragment ion
255.17		255.14	
159.08		159.08	

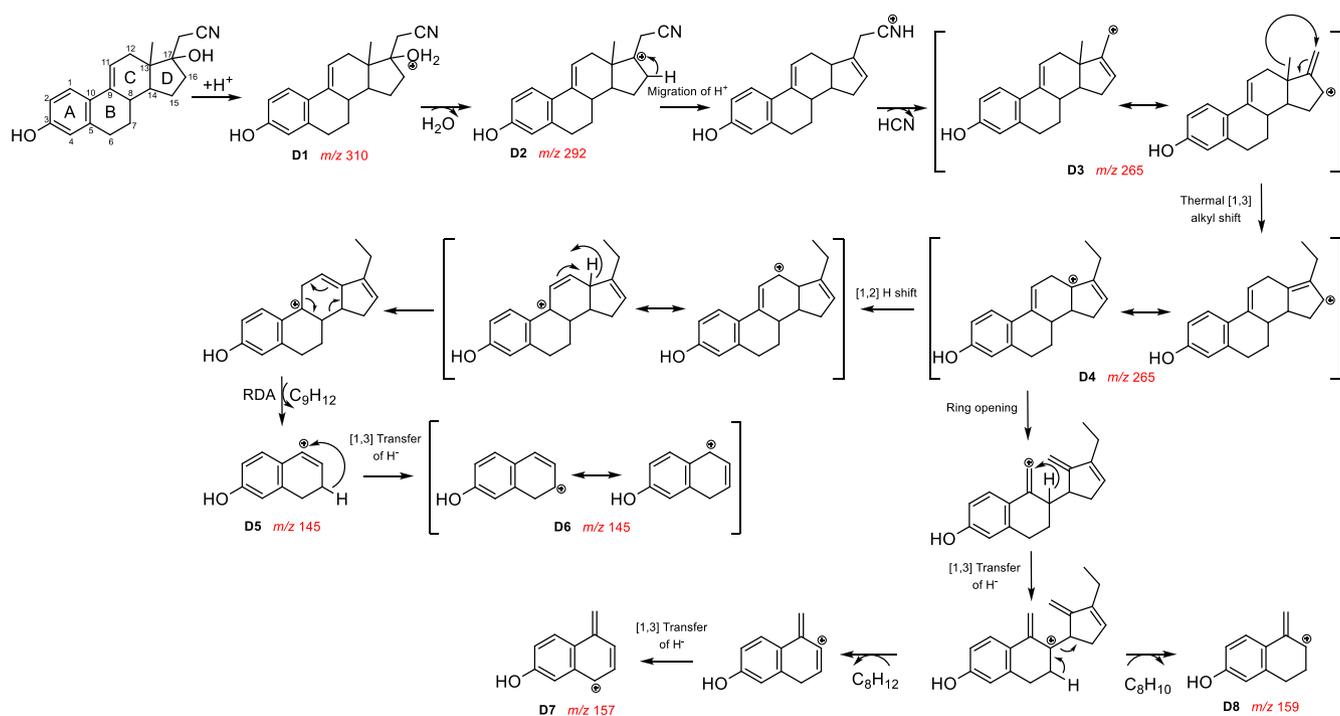
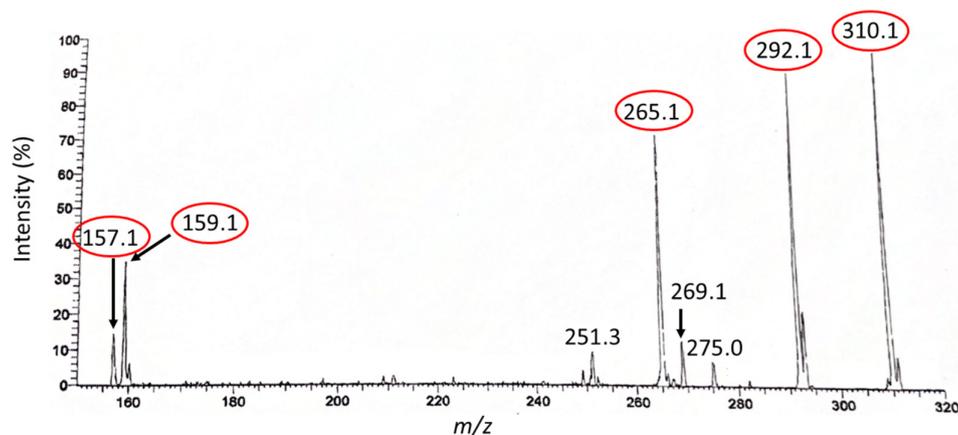


TABLE 3 | ESI-MS² data of **2**, **3**, and **4** under ESI(+) conditions (mass range: m/z 100–320, 150–320, or 60–220).

Substance	Formula	$[M + H]^+$	Fragment ions
			m/z
Estriol 3-methyl ether (2)	C ₁₉ H ₂₆ O ₃	303 (100%)	285 (38%), 274 (8%), 267 (70%), 257 (10%), 241 (16%), 227 (13%), 211 (12%), 199 (10%), 185 (26%), 173 (6%), 171 (11%), 151 (12%), 147 (19%), 135 (14%), 121 (25%)
$\Delta^{9,11}$ -Dehydro-17 α -cyanomethylestradiol (3)	C ₂₀ H ₂₃ NO ₂	310 (100%)	292 (92%), 275 (8%), 269 (12%), 265 (72%), 251 (10%), 159 (35%), 157 (14%)
2-Cyano-2-phenylbutanoic acid ethyl ester (4)	C ₁₃ H ₁₅ NO ₂	218 (47%)	190 (100%), 162 (10%)

**FIGURE 1** | ESI(+)-MS² spectrum of **3**. The precursor ion $[M+H]^+$ at m/z 310 and the marked fragment ions were predicted by ChemFrag; m/z values used for the calculation of the weighted scores and the absolute scores: m/z 310 (100%), 292 (92%), 275 (8%), 269 (12%), 265 (72%), 251 (10%), 159 (35%), 157 (14%).**TABLE 4** | Comparison of the weighted scores and the absolute scores (in brackets) of selected carboxylic acid derivatives, a hydrazine, a thiophosphoric acid ester, sulfonamides, and nitrogen-, sulfur-, and oxygen-containing heterocyclic compounds. The annotation of the (fragment) ions was performed with ChemFrag, MetFrag, and CFM-ID (scores in the format “determined score/maximum score”; peaks with a rel. intensity $\geq 10\%$ [compounds **4** and **6**] or 5% (all other compounds) are taken into account in the evaluation).

Substance	ChemFrag	MetFrag	CFM-ID
<i>N</i> -Ethylnicotinamide	165/165 (4/4)	113/165 (3/4)	50/165 (1/4)
2-Cyano-2-phenylbutanoic acid ethyl ester (4)	153/157 (3/3)	105/157 (2/3)	47/157 (1/3)
2-Cyano-3-methylhexanoic acid ethyl ester	194/214 (4/5)	30/214 (2/5)	83/214 (1/5)
Nicotinamide (5)	195/195 (5/5)	155/195 (3/5)	40/195 (1/5)
2,4-Diamino-6-(hydroxymethyl)pteridine	192/216 (3/4)	185/216 (3/4)	81/216 (1/4)
Gluconic phenylhydrazide	253/260 (8/9)	233/260 (8/9)	23/260 (1/9)
Moxonidine	136/152 (3/5)	121/152 (4/5)	131/152 (2/5)
Hippuric acid methyl ester	203/203 (4/4)	119/203 (3/4)	90/203 (2/4)
<i>p</i> -Toluenesulfonamide	174/346 (4/7)	180/346 (4/7)	180/346 (4/7)
Sulfamethazine	227/291 (5/8)	103/291 (4/8)	203/291 (5/8)
Thiamethoxam	134/281 (4/9)	202/281 (4/9)	91/281 (3/9)
Quercetin (6)	395/395 (7/7)	310/395 (6/7)	45/395 (1/7)
Chlorpyrifos	85/310 (4/9)	135/310 (3/9)	169/310 (4/9)
Total score	2506/3185 (58/79)	1991/3185 (49/79)	1233/3185 (27/79)

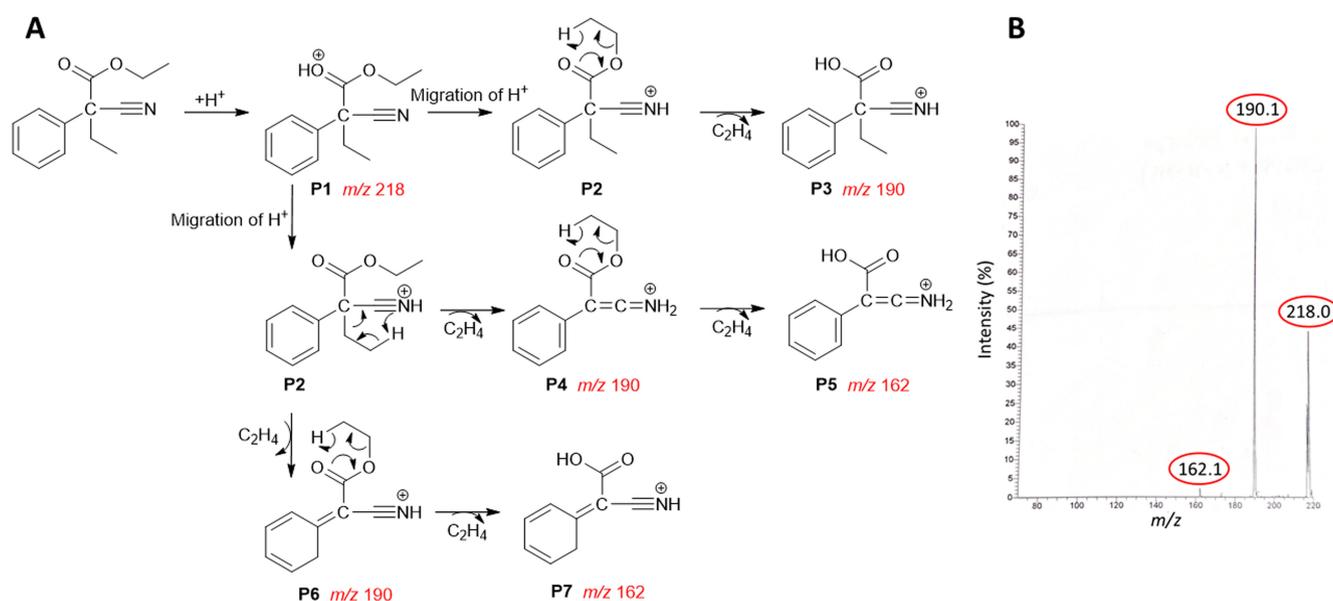
ChemFrag also achieves very good results for the steroids **2** and **3**. Published fragmentation pathways are not yet available for these substances. Consequently, this is the first time that fragmentation pathways for these structures have been predicted on the basis of ChemFrag (steroid **3**: see Scheme 3; steroid **2**: see Scheme S1). Fragment ions of these compounds, which were detected in ESI(+)-MS² analyses, are given in Table 3. In the following, a closer look at the fragmentation pathway of the protonated molecular ion of **3** (Scheme 3, Figure 1) is performed. Compound **3** is protonated at the hydroxy group in position C-17 (**D1**, [C₂₀H₂₄NO₂]⁺, *m/z* 310.18) followed by water elimination (**D2**, [C₂₀H₂₂NO]⁺, *m/z* 292.17). Then, proton transfer and C-CN cleavage occur, leading to the loss of HCN and the formation of an allylic carbocation at *m/z* 265.16 (**D3**, [C₁₉H₂₁O]⁺). Subsequently, a tertiary carbocation (**D4**) is formed by allylic rearrangement and thermal [1,3] alkyl shift. A subsequent rearrangement of a hydrogen atom to an adjacent carbon atom with concurrent rearrangement of the charge results in an allylic carbocation. The fragment ion **D5** at *m/z* 145.07 ([C₁₀H₉O]⁺) is formed by a hydride transfer and a retro-Diels-Alder (RDA) reaction. A further hydride transfer generates the resonantly stabilized fragment ion **D6**. However, the formation of the fragment ion **D6** cannot be proven experimentally, as data are only available for the *m/z* range 150–320. Starting from the fragment ion **D4**, the fragment ions **D7** at *m/z* 157.06 ([C₁₁H₉O]⁺) and **D8** at *m/z* 159.08 ([C₁₁H₁₁O]⁺) can be formed by opening of the C-ring, formation of a tertiary carbocation by hydride transfer and cleavage of the D ring.

3.2 | Application to Other Substance Classes

In addition to the application to steroids, we are able to successfully confirm published fragmentation pathways for various organic compounds and propose new fragmentation pathways using ChemFrag (for structures, see Figure S3 and Schemes S2 and S3). This includes, for instance, the confirmation of the

reaction pathway of **5** according to the studies of Hau et al. [25] and the prediction of a pathway for **4**.

In the first step, the number of annotated (fragment) ions of 13 compounds shown in Figure S3 and Schemes S2 and S3 was determined using ChemFrag, MetFrag, and CFM-ID, again considering the absolute number of annotated peaks and the peak intensities (Table 4). ChemFrag achieves a significantly higher annotation rate for these compounds compared with CFM-ID. A comparison of the total absolute score shows that ChemFrag annotates 58 and MetFrag 49 of the total 79 (fragment) ions for these compounds. In addition, differences can be recognized in the weighted scores. ChemFrag has a total weighted score of 2506/3185, whereas MetFrag has 1991/3185. This difference is probably also related to the missing annotation of the [M+H]⁺ signals when using MetFrag. To evaluate the data, the fragmentation pathways predicted by ChemFrag were considered. For instance, the predicted reaction pathway of the nitrogen-containing heterocyclic compound **5** (Scheme S2) was compared with the reaction pathway published by Hau et al. [25], and it was found that the fragment ions largely coincide. As a further example, the predicted reaction pathway of **4**, a butyric acid derivative, is shown in Scheme 4 (fragmentation path has not yet been published; see Table 3 for ESI-MS² data of **4**). Starting from the protonated molecular ion of **4** (**P1**, [C₁₃H₁₆NO₂]⁺, *m/z* 218.12), ion **P2** is formed by a proton shift to the -C≡N group. In the next step, cleavage of the Alk-O bond leads to the formation of the acid **P3** at *m/z* 190.09 ([C₁₁H₁₂NO₂]⁺) and the corresponding alkene C₂H₄. Starting from **P2**, an alkene elimination from the acid side of the ester via a McLafferty rearrangement involving the -C≡N group also generates a fragment ion at *m/z* 190.09 (**P4**, [C₁₁H₁₂NO₂]⁺). A subsequent H rearrangement on the alcohol side of the ester and the elimination of the alkene C₂H₄ leads to the acid **P5** ([C₉H₈NO₂]⁺, *m/z* 162.06). A McLafferty rearrangement involving the benzene ring with C₂H₄ cleavage leads to the formation of the fragment ion **P6** at *m/z* 190.09 ([C₁₁



SCHEME 4 | (A) Fragmentation pathway of protonated **4** [M+H]⁺ predicted by ChemFrag. (B) ESI(+)-MS² spectrum of **4**. The precursor ion [M+H]⁺ at *m/z* 218 and the marked fragment ions were predicted by ChemFrag.

H₁₂NO₂⁺). A further C₂H₄ cleavage leads to the fragment ion **P7** at *m/z* 162.06 ([C₉H₈NO₂]⁺).

In addition, the fragmentation pathway of the flavonoid **6** was predicted as an example of an oxygen-containing heterocyclic compound (Scheme S3). The predicted reaction pathway is largely consistent with the studies by Tsimogiannis et al. or Burgert [34, 35].

4 | Conclusion

ChemFrag has been successfully used for the interpretation of ESI(+)-MSⁿ fragmentation spectra of antibiotics, pesticides, natural products and structural analogs such as estradiol derivatives. The cleavage and rearrangement rules, implemented in this study, extend the scope of ChemFrag and enable, for example, the annotation of fragment ions of steroids. A comparison with fragmentation pathways published in other studies has shown that ChemFrag provides reliable annotations for compounds such as **1** or **5**. The number of annotated (fragment) ions is comparable or higher than that of the established programs MetFrag and CFM-ID, whereby fragment ions with higher intensity in particular are annotated. Furthermore, using the example of compound **1**, it was shown that ChemFrag predicts chemically more meaningful annotations than CFM-ID in some cases. Thus, the combined approach of ChemFrag, using quantum chemistry as well as rule-based fragmentation, proves to be a valuable addition to established programs.

Acknowledgments

The authors are grateful to late Dr. R. Kluge (Martin-Luther-Universität Halle-Wittenberg) for recording the ESI-MSⁿ spectra. Dr. A. E. Kramell is grateful for financial support from the German Research Foundation (DFG, project number 425225219). Open Access funding enabled and organized by Projekt DEAL.

Data Availability Statement

The data that support the findings of this study are available from the corresponding author upon reasonable request.

References

1. D. H. Nguyen, C. H. Nguyen, and H. Mamitsuka, "Recent Advances and Prospects of Computational Methods for Metabolite Identification: A Review With Emphasis on Machine Learning Approaches," *Briefings in Bioinformatics* 20, no. 6 (2018): 2028–2043, <https://doi.org/10.1093/bib/bby066>.
2. F. Hufsky, K. Scheubert, and S. Böcker, "Computational Mass Spectrometry for Small-Molecule Fragmentation," *Trends in Analytical Chemistry* 53 (2014): 41–48, <https://doi.org/10.1016/j.trac.2013.09.008>.
3. E. L. Schymanski, M. Meringer, and W. Brack, "Matching Structures to Mass Spectra Using Fragmentation Patterns: Are the Results as Good as They Look?," *Analytical Chemistry* 81, no. 9 (2009): 3608–3617, <https://doi.org/10.1021/ac802715e>.
4. A. Hill and R. J. Mortishire-Smith, "Automated Assignment of High-Resolution Collisionally Activated Dissociation Mass Spectra Using a Systematic Bond Disconnection Approach," *Rapid Communications in Mass Spectrometry* 19 (2005): 3111–3118, <https://doi.org/10.1002/rcm.2177>.
5. M. Heinonen, A. Rantanen, T. Mielikäinen, et al., "FiD: A Software for Ab Initio Structural Identification of Product Ions From Tandem Mass Spectrometric Data," *Rapid Communications in Mass Spectrometry* 22, no. 19 (2008): 3043–3052, <https://doi.org/10.1002/rcm.3701>.
6. F. Rasche, K. Scheubert, F. Hufsky, et al., "Identifying the Unknowns by Aligning Fragmentation Trees," *Analytical Chemistry* 84, no. 7 (2012): 3417–3426, <https://doi.org/10.1021/ac300304u>.
7. F. Rasche, A. Svatoš, R. K. Maddula, C. Böttcher, and S. Böcker, "Computing Fragmentation Trees From Tandem Mass Spectrometry Data," *Analytical Chemistry* 83, no. 4 (2011): 1243–1251, <https://doi.org/10.1021/ac101825k>.
8. M. Heinonen, H. Shen, N. Zamboni, and J. Rousu, "Metabolite Identification and Molecular Fingerprint Prediction Through Machine Learning," *Bioinformatics* 28, no. 18 (2012): 2333–2341, <https://doi.org/10.1093/bioinformatics/bts437>.
9. K. Dührkop, H. Shen, M. Meusel, J. Rousu, and S. Böcker, "Searching Molecular Structure Databases With Tandem Mass Spectra Using CSI:FingerID," *Proceedings of the National Academy of Sciences of the United States of America* 112, no. 41 (2015): 12580–12585, <https://doi.org/10.1073/pnas.1509788112>.
10. Y. Djoumbou-Feunang, A. Pon, N. Karu, et al., "CFM-ID 3.0: Significantly Improved ESI-MS/MS Prediction and Compound Identification," *Metabolites* 9, no. 4 (2019): 72, <https://doi.org/10.3390/metabo9040072>.
11. C. Ruttkies, E. L. Schymanski, S. Wolf, J. Hollender, and S. Neumann, "MetFrag Relunched: Incorporating Strategies Beyond In Silico Fragmentation," *Journal of Cheminformatics* 8, no. 1 (2016): 3, <https://doi.org/10.1186/s13321-016-0115-9>.
12. F. Wang, D. Allen, S. Tian, et al., "CFM-ID 4.0—A Web Server for Accurate MS-Based Metabolite Identification," *Nucleic Acids Research* 50, no. W1 (2022): W165–W174, <https://doi.org/10.1093/nar/gkac383>.
13. K. Dührkop, M. Fleischauer, M. Ludwig, et al., "SIRIUS 4: A Rapid Tool for Turning Tandem Mass Spectra Into Metabolite Structure Information," *Nature Methods* 16, no. 4 (2019): 299–302, <https://doi.org/10.1038/s41592-019-0344-8>.
14. S. Grimme, "Towards First Principles Calculation of Electron Impact Mass Spectra of Molecules," *Angewandte Chemie, International Edition* 52, no. 24 (2013): 6306–6312, <https://doi.org/10.1002/anie.201300158>.
15. J. Cautereels, M. Claeys, D. Geldof, and F. Blockhuys, "Quantum Chemical Mass Spectrometry: Ab Initio Prediction of Electron Ionization Mass Spectra and Identification of New Fragmentation Pathways," *Journal of Mass Spectrometry* 51, no. 8 (2016): 602–614, <https://doi.org/10.1002/jms.3791>.
16. J. Cautereels and F. Blockhuys, "Quantum Chemical Mass Spectrometry: Verification and Extension of the Mobile Proton Model for Histidine," *Journal of the American Society for Mass Spectrometry* 28, no. 6 (2017): 1227–1235, <https://doi.org/10.1007/s13361-017-1636-9>.
17. B. G. Janesko, L. Li, and R. Mensing, "Quantum Chemical Fragment Precursor Tests: Accelerating De Novo Annotation of Tandem Mass Spectra," *Analytica Chimica Acta* 995 (2017): 52–64, <https://doi.org/10.1016/j.aca.2017.09.034>.
18. J.-A. Schüler, S. Neumann, M. Müller-Hannemann, and W. Brandt, "ChemFrag: Chemically Meaningful Annotation of Fragment Ion Mass Spectra," *Journal of Mass Spectrometry* 53, no. 11 (2018): 1104–1115, <https://doi.org/10.1002/jms.4278>.
19. D. P. Demarque, A. E. M. Crotti, R. Vessecchi, J. L. C. Lopes, and N. P. Lopes, "Fragmentation Reactions Using Electrospray Ionization Mass Spectrometry: An Important Tool for the Structural Elucidation and Characterization of Synthetic and Natural Products," *Natural Product Reports* 33, no. 3 (2016): 432–455, <https://doi.org/10.1039/c5np00073d>.
20. L. Ma and S. R. Yates, "A Review on Structural Elucidation of Metabolites of Environmental Steroid Hormones via Liquid

- Chromatography–Mass Spectrometry,” *Trends in Analytical Chemistry* 109 (2018): 142–153, <https://doi.org/10.1016/j.trac.2018.10.007>.
21. A. Weissberg and S. Dagan, “Interpretation of ESI(+)-MS-MS Spectra—Towards the Identification of “Unknowns”,” *International Journal of Mass Spectrometry* 299, no. 2 (2011): 158–168, <https://doi.org/10.1016/j.ijms.2010.10.024>.
22. E. Pretsch, P. Bühlmann, and M. Badertscher, “Mass Spectrometry,” in *Structure Determination of Organic Compounds: Tables of Spectral Data*, eds. E. Pretsch, P. Bühlmann, and M. Badertscher (Springer Berlin Heidelberg, 2020), 375–443.
23. J. H. Gross, “Fragmentierungsreaktionen,” in *Massenspektrometrie: Spektroskopiekurs Kompakt*, ed. J. H. Gross (Springer Berlin Heidelberg, 2019), 33–56.
24. M. S. Refat, S. A. El-Korashy, and A. S. Ahmed, “Synthesis and Characterization of Mn(II), Au(III) and Zr(IV) Hippurates Complexes,” *Spectrochimica Acta. Part A, Molecular and Biomolecular Spectroscopy* 70, no. 4 (2008): 840–849, <https://doi.org/10.1016/j.saa.2007.09.020>.
25. J. Hau, R. Stadler, T. A. Jenny, and L. B. Fay, “Tandem Mass Spectrometric Accurate Mass Performance of Time-of-Flight and Fourier Transform Ion Cyclotron Resonance Mass Spectrometry: A Case Study With Pyridine Derivatives,” *Rapid Communications in Mass Spectrometry* 15, no. 19 (2001): 1840–1848, <https://doi.org/10.1002/rcm.444>.
26. J. B. Bialecki, C. S. Weisbecker, and A. B. Attygalle, “Low-Energy Collision-Induced Dissociation Mass Spectra of Protonated *p*-Toluenesulfonamides Derived From Aliphatic Amines,” *Journal of the American Society for Mass Spectrometry* 25, no. 6 (2014): 1068–1078, <https://doi.org/10.1007/s13361-014-0865-4>.
27. S. Rafqah, Z. S. Seddigi, S. A. Ahmed, E. Danish, and M. Sarakha, “Use of Quadrupole Time of Flight Mass Spectrometry for the Characterization of Transformation Products of the Antibiotic Sulfamethazine Upon Photocatalysis With Pd-Doped Ceria-ZnO Nanocomposite,” *Journal of Mass Spectrometry* 50, no. 2 (2015): 298–307, <https://doi.org/10.1002/jms.3521>.
28. C. Bannwarth, S. Ehlert, and S. Grimme, “GFN2-xTB—An Accurate and Broadly Parametrized Self-Consistent Tight-Binding Quantum Chemical Method With Multipole Electrostatics and Density-Dependent Dispersion Contributions,” *Journal of Chemical Theory and Computation* 15, no. 3 (2019): 1652–1671, <https://doi.org/10.1021/acs.jctc.8b01176>.
29. M. Thevis, U. Bommerich, G. Opfermann, and W. Schänzer, “Characterization of Chemically Modified Steroids for Doping Control Purposes by Electrospray Ionization Tandem Mass Spectrometry,” *Journal of Mass Spectrometry* 40, no. 4 (2005): 494–502, <https://doi.org/10.1002/jms.820>.
30. W. F. Johns, “Retropinacol Rearrangement of Estradiol 3-Methyl Ether,” *Journal of Organic Chemistry* 26, no. 11 (1961): 4583–4591, <https://doi.org/10.1021/jo01069a091>.
31. S. Bourcier, C. Poisson, Y. Souissi, S. Kinani, S. Bouchonnet, and M. Sablier, “Elucidation of the Decomposition Pathways of Protonated and Deprotonated Estrone Ions: Application to the Identification of Photolysis Products,” *Rapid Communications in Mass Spectrometry* 24, no. 20 (2010): 2999–3010, <https://doi.org/10.1002/rcm.4722>.
32. F. Guan, L. R. Soma, Y. Luo, C. E. Uboh, and S. Peterman, “Collision-Induced Dissociation Pathways of Anabolic Steroids by Electrospray Ionization Tandem Mass Spectrometry,” *Journal of the American Society for Mass Spectrometry* 17, no. 4 (2006): 477–489, <https://doi.org/10.1016/j.jasms.2005.11.021>.
33. MassBank/MassBank-Data: Release Version 2024. Zenodo; 2024.
34. D. Tsimogiannis, M. Samiotaki, G. Panayotou, and V. Oreopoulou, “Characterization of Flavonoid Subgroups and Hydroxy Substitution by HPLC-MS/MS,” *Molecules* 12, no. 3 (2007): 593–606, <https://doi.org/10.3390/12030593>.
35. M. Burgert, *Aufbau Einer Flavonoid MSⁿ Datenbank Mittels ESI-Ionenfallen-Massenspektrometrie* (Pharmacognosy, Universität Wien, 2011).

Supporting Information

Additional supporting information can be found online in the Supporting Information section.